

# Bis(«beta»-chloroethyl) sulfoxide

<b>Other names:</b>	1,1'-Sulfinylbis[2-chloroethane] 1-chloro-2-(2-chloroethylsulfinyl)ethane Bis(2-chloroethyl) sulfoxide Ethane, 1,1'-sulfinylbis[2-chloro- H Sulfoxide Mustard gas sulfoxide Mustard sulfone NSC 141393 Sulfoxide, bis(2-chloroethyl) Yperite sulfoxide
<b>Inchi:</b>	InChI=1S/C4H8Cl2OS/c5-1-3-8(7)4-2-6/h1-4H2
<b>InchiKey:</b>	NOMHBBFEJSVGSC-UHFFFAOYSA-N
<b>Formula:</b>	C4H8Cl2OS
<b>SMILES:</b>	O=S(CCCI)CCCI
<b>Mol. weight [g/mol]:</b>	175.08
<b>CAS:</b>	5819-08-9

## Physical Properties

Property code	Value	Unit	Source
gf	-258.77	kJ/mol	Joback Method
hf	-363.11	kJ/mol	Joback Method
hfus	22.26	kJ/mol	Joback Method
hvap	45.99	kJ/mol	Joback Method
log10ws	-1.16		Aqueous Solubility Prediction Method
logp	1.213		Crippen Method
mcvol	113.920	ml/mol	McGowan Method
pc	3955.54	kPa	Joback Method
tb	424.06	K	Joback Method
tc	616.88	K	Joback Method
tf	231.16	K	Joback Method
vc	0.448	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	193.92	J/mol×K	424.06	Joback Method
cpg	202.17	J/mol×K	456.20	Joback Method
cpg	210.07	J/mol×K	488.33	Joback Method
cpg	217.62	J/mol×K	520.47	Joback Method
cpg	224.82	J/mol×K	552.61	Joback Method
cpg	231.68	J/mol×K	584.75	Joback Method
cpg	238.20	J/mol×K	616.88	Joback Method

## Sources

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C5819089&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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