

# Ethane, 1,2-dibromo-1-ethoxy-

<b>Other names:</b>	1,2-dibromo-1-ethoxyethane
<b>Inchi:</b>	InChI=1S/C4H8Br2O/c1-2-7-4(6)3-5/h4H,2-3H2,1H3
<b>InchiKey:</b>	YNQUMIFEHUKSMO-UHFFFAOYSA-N
<b>Formula:</b>	C4H8Br2O
<b>SMILES:</b>	CCOC(Br)CBr
<b>Mol. weight [g/mol]:</b>	231.91
<b>CAS:</b>	2983-26-8

## Physical Properties

Property code	Value	Unit	Source
gf	-96.00	kJ/mol	Joback Method
hf	-210.73	kJ/mol	Joback Method
hfus	14.35	kJ/mol	Joback Method
hvap	39.39	kJ/mol	Joback Method
log10ws	-2.05		Crippen Method
logp	2.139		Crippen Method
mcvol	108.090	ml/mol	McGowan Method
pc	4608.87	kPa	Joback Method
rinpol	990.00		NIST Webbook
tb	445.22	K	Joback Method
tc	655.40	K	Joback Method
tf	261.67	K	Joback Method
vc	0.396	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	177.84	J/molxK	445.22	Joback Method
cpg	213.13	J/molxK	620.37	Joback Method
cpg	206.79	J/molxK	585.34	Joback Method
cpg	200.11	J/molxK	550.31	Joback Method
cpg	193.06	J/molxK	515.28	Joback Method
cpg	185.64	J/molxK	480.25	Joback Method
cpg	219.12	J/molxK	655.40	Joback Method

dvisc	0.0003471	Paxs	445.22	Joback Method
dvisc	0.0004389	Paxs	414.63	Joback Method
dvisc	0.0005762	Paxs	384.04	Joback Method
dvisc	0.0007929	Paxs	353.45	Joback Method
dvisc	0.0011592	Paxs	322.85	Joback Method
dvisc	0.0018349	Paxs	292.26	Joback Method
dvisc	0.0032337	Paxs	261.67	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2983268&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2983268&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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