

# Ethyl methylaminodithiocarbamate

<b>Inchi:</b>	InChI=1S/C4H9NS2/c1-3-7-4(6)5-2/h3H2,1-2H3,(H,5,6)
<b>InchiKey:</b>	MNKSXVVWOZQGLR-UHFFFAOYSA-N
<b>Formula:</b>	C4H9NS2
<b>SMILES:</b>	CCSC(S)=NC
<b>Mol. weight [g/mol]:</b>	135.25

## Physical Properties

Property code	Value	Unit	Source
hf	26.89	kJ/mol	Joback Method
hvap	41.45	kJ/mol	Joback Method
log10ws	-1.61		Crippen Method
logp	1.655		Crippen Method
mcvol	105.600	ml/mol	McGowan Method
pc	3805.69	kPa	Joback Method
rinpol	1479.00		NIST Webbook
rinpol	1450.00		NIST Webbook
rinpol	1454.00		NIST Webbook
rinpol	1452.00		NIST Webbook
rinpol	1465.00		NIST Webbook
rinpol	1472.00		NIST Webbook
rinpol	1450.00		NIST Webbook
rinpol	1479.00		NIST Webbook
tb	499.12	K	Joback Method
tc	745.00	K	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R11200&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R11200&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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