

# Hydroxyproline

<b>Other names:</b>	(2S,4R)-4-hydroxypyrrolidine-2-carboxylic acid 4-Hydroxy-2-pyrrolidinecarboxylic acid 4-Hydroxy-L-proline 4-Hydroxyproline 4-L-Hydroxyproline Hydroxy-L-Proline Hydroxyproline, (L)- Hypro L-4-Hydroxyproline L-Hydroxyproline L-Proline, 4-hydroxy-, (4R)- L-Proline, 4-hydroxy-, trans- Ls-Hydroxyproline NSC 46704 Proline, 4-hydroxy- Proline, 4-hydroxy-, L- trans-4-Hydroxy-L-Proline trans-4-Hydroxyproline trans-Hydroxyproline trans-L-4-hydroxyproline trans-L-Hydroxyproline «delta»-Hydroxyproline
<b>Inchi:</b>	InChI=1S/C5H9NO3/c7-3-1-4(5(8)9)6-2-3/h3-4,6-7H,1-2H2,(H,8,9)/t3?,4-/m1/s1
<b>InchiKey:</b>	PMMYEEVYMWASQN-SRBOSORUSA-N
<b>Formula:</b>	C5H9NO3
<b>SMILES:</b>	O=C(O)C1CC(O)CN1
<b>Mol. weight [g/mol]:</b>	131.13
<b>CAS:</b>	51-35-4

## Physical Properties

Property code	Value	Unit	Source
gf	-294.79	kJ/mol	Joback Method
hf	-485.62	kJ/mol	Joback Method
hfus	23.08	kJ/mol	Joback Method
hvap	73.53	kJ/mol	Joback Method
log10ws	0.41		Crippen Method

logp	-1.206		Crippen Method
mvol	93.740	ml/mol	McGowan Method
pc	6113.06	kPa	Joback Method
tb	611.19	K	Joback Method
tc	804.48	K	Joback Method
tf	429.37	K	Joback Method
vc	0.337	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	247.66	J/mol×K	611.19	Joback Method
cpg	256.55	J/mol×K	643.40	Joback Method
cpg	264.97	J/mol×K	675.62	Joback Method
cpg	272.92	J/mol×K	707.83	Joback Method
cpg	280.40	J/mol×K	740.05	Joback Method
cpg	287.43	J/mol×K	772.26	Joback Method
cpg	294.02	J/mol×K	804.48	Joback Method

## Sources

Thermodynamic properties of aqueous 4-hydroxyproline at different temperatures:	<a href="https://www.doi.org/10.1016/j.jct.2009.11.017">https://www.doi.org/10.1016/j.jct.2009.11.017</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C51354&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C51354&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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