

# 1,7-Difluoro-1,1,3,5,7,7-hexanitro-3,5-diazaheptane

<b>Inchi:</b>	InChI=1S/C5H6F2N8O12/c6-4(10(16)17,11(18)19)1-8(14(24)25)3-9(15(26)27)2-5(7,12(28)29)
<b>InchiKey:</b>	SGOLSSPCBCZHEA-UHFFFAOYSA-N
<b>Formula:</b>	C5H6F2N8O12
<b>SMILES:</b>	O=[N+](=[O-])N(CN(CC(F)([N+](=O)[O-])[N+](=O)[O-])[N+](=O)[O-])CC(F)([N+](=O)[O-])[N+](=O)[O-]
<b>Mol. weight [g/mol]:</b>	408.14
<b>CAS:</b>	80308-93-6

## Physical Properties

Property code	Value	Unit	Source
gf	42.14	kJ/mol	Joback Method
hf	-485.75	kJ/mol	Joback Method
hfus	74.25	kJ/mol	Joback Method
hvap	126.13	kJ/mol	Joback Method
log10ws	-3.95		Crippen Method
logp	-1.715		Crippen Method
mcvol	209.330	ml/mol	McGowan Method
pc	3534.66	kPa	Joback Method
tb	1241.80	K	Joback Method
tc	1537.01	K	Joback Method
tf	1078.73	K	Joback Method
vc	0.858	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	681.69	J/molxK	1241.80	Joback Method
cpg	689.76	J/molxK	1291.00	Joback Method
cpg	698.88	J/molxK	1340.20	Joback Method
cpg	709.34	J/molxK	1389.40	Joback Method
cpg	721.41	J/molxK	1438.61	Joback Method
cpg	735.38	J/molxK	1487.81	Joback Method
cpg	751.54	J/molxK	1537.01	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C80308936&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C80308936&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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