

# 1,2-Dichloropropyl propyl ether

<b>Inchi:</b>	InChI=1S/C6H12Cl2O/c1-3-4-9-6(8)5(2)7/h5-6H,3-4H2,1-2H3
<b>InchiKey:</b>	BGZOWNBTBBZBLL-UHFFFAOYSA-N
<b>Formula:</b>	C6H12Cl2O
<b>SMILES:</b>	CCCOC(Cl)C(C)Cl
<b>Mol. weight [g/mol]:</b>	171.06

## Physical Properties

Property code	Value	Unit	Source
gf	-134.10	kJ/mol	Joback Method
hf	-341.43	kJ/mol	Joback Method
hfus	13.83	kJ/mol	Joback Method
hvap	39.35	kJ/mol	Joback Method
log10ws	-2.45		Crippen Method
logp	2.605		Crippen Method
mcvol	125.750	ml/mol	McGowan Method
pc	2862.74	kPa	Joback Method
tb	433.08	K	Joback Method
tc	621.21	K	Joback Method
tf	209.45	K	Joback Method
vc	0.475	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	234.53	J/molxK	433.08	Joback Method
cpg	245.04	J/molxK	464.43	Joback Method
cpg	255.15	J/molxK	495.79	Joback Method
cpg	264.85	J/molxK	527.14	Joback Method
cpg	274.14	J/molxK	558.50	Joback Method
cpg	283.05	J/molxK	589.85	Joback Method
cpg	291.56	J/molxK	621.21	Joback Method
dvisc	0.0084256	Paxs	209.45	Joback Method
dvisc	0.0030348	Paxs	246.72	Joback Method
dvisc	0.0014291	Paxs	283.99	Joback Method

dvisc	0.0008015	Paxs	321.26	Joback Method
dvisc	0.0005069	Paxs	358.54	Joback Method
dvisc	0.0003495	Paxs	395.81	Joback Method
dvisc	0.0002569	Paxs	433.08	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R629073&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R629073&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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