

# 1,1-Dichloropropyl propyl ether

<b>Inchi:</b>	InChI=1S/C6H12Cl2O/c1-3-5-9-6(7,8)4-2/h3-5H2,1-2H3
<b>InchiKey:</b>	UFLXLVHVFC SXIR-UHFFFAOYSA-N
<b>Formula:</b>	C6H12Cl2O
<b>SMILES:</b>	CCCOC(Cl)(Cl)CC
<b>Mol. weight [g/mol]:</b>	171.06

## Physical Properties

Property code	Value	Unit	Source
gf	-126.38	kJ/mol	Joback Method
hf	-339.62	kJ/mol	Joback Method
hfus	13.46	kJ/mol	Joback Method
hvap	38.83	kJ/mol	Joback Method
log10ws	-2.83		Crippen Method
logp	2.954		Crippen Method
mvol	125.750	ml/mol	McGowan Method
pc	2871.95	kPa	Joback Method
rinpol	954.00		NIST Webbook
rinpol	954.00		NIST Webbook
tb	430.73	K	Joback Method
tc	622.33	K	Joback Method
tf	241.87	K	Joback Method
vc	0.476	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	236.86	J/molxK	430.73	Joback Method
cpg	286.74	J/molxK	590.40	Joback Method
cpg	277.80	J/molxK	558.46	Joback Method
cpg	268.36	J/molxK	526.53	Joback Method
cpg	258.40	J/molxK	494.60	Joback Method
cpg	247.91	J/molxK	462.66	Joback Method
cpg	295.21	J/molxK	622.33	Joback Method
dvisc	0.0002939	Paxs	430.73	Joback Method

dvisc	0.0003925	Paxs	399.25	Joback Method
dvisc	0.0005509	Paxs	367.78	Joback Method
dvisc	0.0008236	Paxs	336.30	Joback Method
dvisc	0.0013382	Paxs	304.82	Joback Method
dvisc	0.0024314	Paxs	273.35	Joback Method
dvisc	0.0051604	Paxs	241.87	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R629068&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R629068&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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