

# Butanamide, 2-bromo-2-ethyl-

<b>Other names:</b>	Butyramide, 2-bromo-2-ethyl- Carabromide Carbromide Diethylbromoacetamide Neuronal 2-Bromo-2-ethylbutyramide 2-Bromo-2,2-diethylacetamide 2-Ethyl-2-bromobutyramide 2-Bromo-2-ethylbutanamide 2-Bromo-2-ethylbutylamide NSC 4607
<b>Inchi:</b>	InChI=1S/C6H12BrNO/c1-3-6(7,4-2)5(8)9/h3-4H2,1-2H3,(H2,8,9)
<b>InchiKey:</b>	ACEYAMOAGUDVAQ-UHFFFAOYSA-N
<b>Formula:</b>	C6H12BrNO
<b>SMILES:</b>	CCC(Br)(CC)C(=N)O
<b>Mol. weight [g/mol]:</b>	194.07
<b>CAS:</b>	511-70-6

## Physical Properties

Property code	Value	Unit	Source
gf	83.58	kJ/mol	Joback Method
hf	-103.49	kJ/mol	Joback Method
hvap	62.85	kJ/mol	Joback Method
log10ws	-3.61		Crippen Method
logp	2.475		Crippen Method
mcvol	124.450	ml/mol	McGowan Method
rinpol	1215.00		NIST Webbook
rinpol	1192.00		NIST Webbook
rinpol	1215.00		NIST Webbook
rinpol	1215.00		NIST Webbook
rinpol	1192.00		NIST Webbook
rinpol	1205.00		NIST Webbook
tb	576.13	K	Joback Method
tf	349.20	K	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	282.39	J/mol×K	576.13	Joback Method
cpg	46.33	J/mol×K	100.12	Joback Method
cpg	46.33	J/mol×K	100.12	Joback Method
cpg	46.33	J/mol×K	100.12	Joback Method
cpg	46.33	J/mol×K	100.12	Joback Method
cpg	46.33	J/mol×K	100.12	Joback Method
cpg	46.33	J/mol×K	100.12	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C511706&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C511706&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tf:</b>	Normal melting (fusion) point

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