

# 3-Pentanone, 2,4-dichloro-2-methyl

<b>Inchi:</b>	InChI=1S/C6H10Cl2O/c1-4(7)5(9)6(2,3)8/h4H,1-3H3
<b>InchiKey:</b>	OLIGKRNLJKEAV-UHFFFAOYSA-N
<b>Formula:</b>	C6H10Cl2O
<b>SMILES:</b>	CC(Cl)C(=O)C(C)(C)Cl
<b>Mol. weight [g/mol]:</b>	169.05

## Physical Properties

Property code	Value	Unit	Source
gf	-152.74	kJ/mol	Joback Method
hf	-325.26	kJ/mol	Joback Method
hfus	10.35	kJ/mol	Joback Method
hvap	42.78	kJ/mol	Joback Method
log10ws	-2.14		Crippen Method
logp	2.200		Crippen Method
mcvol	121.450	ml/mol	McGowan Method
pc	3163.27	kPa	Joback Method
rinpol	942.00		NIST Webbook
rinpol	942.00		NIST Webbook
tb	461.74	K	Joback Method
tc	670.58	K	Joback Method
tf	254.57	K	Joback Method
vc	0.459	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	230.86	J/molxK	461.74	Joback Method
cpg	241.62	J/molxK	496.55	Joback Method
cpg	251.67	J/molxK	531.35	Joback Method
cpg	261.06	J/molxK	566.16	Joback Method
cpg	269.82	J/molxK	600.97	Joback Method
cpg	277.98	J/molxK	635.77	Joback Method
cpg	285.58	J/molxK	670.58	Joback Method
dvisc	0.0076920	Paxs	254.57	Joback Method

dvisc	0.0033961	Paxs	289.10	Joback Method
dvisc	0.0017852	Paxs	323.63	Joback Method
dvisc	0.0010623	Paxs	358.15	Joback Method
dvisc	0.0006925	Paxs	392.68	Joback Method
dvisc	0.0004838	Paxs	427.21	Joback Method
dvisc	0.0003566	Paxs	461.74	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R630350&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R630350&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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