

# 3-Hexanone, 2,2-dichloro

<b>Inchi:</b>	InChI=1S/C6H10Cl2O/c1-3-4-5(9)6(2,7)8/h3-4H2,1-2H3
<b>InchiKey:</b>	FFZAXGOQIDURKS-UHFFFAOYSA-N
<b>Formula:</b>	C6H10Cl2O
<b>SMILES:</b>	CCCC(=O)C(C)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	169.05

## Physical Properties

Property code	Value	Unit	Source
gf	-150.30	kJ/mol	Joback Method
hf	-319.98	kJ/mol	Joback Method
hfus	13.88	kJ/mol	Joback Method
hvap	43.17	kJ/mol	Joback Method
log10ws	-2.53		Crippen Method
logp	2.549		Crippen Method
mcvol	121.450	ml/mol	McGowan Method
pc	3135.00	kPa	Joback Method
rinpola	957.00		NIST Webbook
rinpola	957.00		NIST Webbook
tb	462.18	K	Joback Method
tc	666.21	K	Joback Method
tf	269.57	K	Joback Method
vc	0.465	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	230.68	J/mol×K	462.18	Joback Method
cpg	241.10	J/mol×K	496.18	Joback Method
cpg	250.87	J/mol×K	530.19	Joback Method
cpg	260.00	J/mol×K	564.19	Joback Method
cpg	268.54	J/mol×K	598.20	Joback Method
cpg	276.51	J/mol×K	632.20	Joback Method
cpg	283.96	J/mol×K	666.21	Joback Method
dvisc	0.0052226	Paxs	269.57	Joback Method

dvisc	0.0026645	Paxs	301.67	Joback Method
dvisc	0.0015473	Paxs	333.77	Joback Method
dvisc	0.0009885	Paxs	365.88	Joback Method
dvisc	0.0006788	Paxs	397.98	Joback Method
dvisc	0.0004930	Paxs	430.08	Joback Method
dvisc	0.0003744	Paxs	462.18	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R630172&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R630172&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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