

# 2-Pentanone, 1,1-dichloro-4-methyl

<b>Inchi:</b>	InChI=1S/C6H10Cl2O/c1-4(2)3-5(9)6(7)8/h4,6H,3H2,1-2H3
<b>InchiKey:</b>	UBXSUXLQWXOKLD-UHFFFAOYSA-N
<b>Formula:</b>	C6H10Cl2O
<b>SMILES:</b>	CC(C)CC(=O)C(Cl)Cl
<b>Mol. weight [g/mol]:</b>	169.05

## Physical Properties

Property code	Value	Unit	Source
gf	-158.02	kJ/mol	Joback Method
hf	-321.79	kJ/mol	Joback Method
hfus	14.24	kJ/mol	Joback Method
hvap	43.69	kJ/mol	Joback Method
log10ws	-2.29		Crippen Method
logp	2.405		Crippen Method
mcvol	121.450	ml/mol	McGowan Method
pc	3124.49	kPa	Joback Method
rinpol	966.00		NIST Webbook
rinpol	966.00		NIST Webbook
tb	464.53	K	Joback Method
tc	664.81	K	Joback Method
tf	237.15	K	Joback Method
vc	0.464	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	227.95	J/mol×K	464.53	Joback Method
cpg	237.97	J/mol×K	497.91	Joback Method
cpg	247.47	J/mol×K	531.29	Joback Method
cpg	256.47	J/mol×K	564.67	Joback Method
cpg	265.00	J/mol×K	598.05	Joback Method
cpg	273.06	J/mol×K	631.43	Joback Method
cpg	280.67	J/mol×K	664.81	Joback Method
dvisc	0.0080703	Paxs	237.15	Joback Method

dvisc	0.0032987	Paxs	275.05	Joback Method
dvisc	0.0016746	Paxs	312.94	Joback Method
dvisc	0.0009842	Paxs	350.84	Joback Method
dvisc	0.0006416	Paxs	388.74	Joback Method
dvisc	0.0004513	Paxs	426.63	Joback Method
dvisc	0.0003362	Paxs	464.53	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R629944&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R629944&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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