

# Propionic acid, 3-iodo-, propyl ester

<b>Inchi:</b>	InChI=1S/C6H11IO2/c1-2-5-9-6(8)3-4-7/h2-5H2,1H3
<b>InchiKey:</b>	CNTHQHNTVHYYJW-UHFFFAOYSA-N
<b>Formula:</b>	C6H11IO2
<b>SMILES:</b>	CCCOC(=O)CCI
<b>Mol. weight [g/mol]:</b>	242.05

## Physical Properties

Property code	Value	Unit	Source
gf	-176.16	kJ/mol	Joback Method
hf	-335.10	kJ/mol	Joback Method
hfus	18.49	kJ/mol	Joback Method
hvap	47.48	kJ/mol	Joback Method
log10ws	-2.14		Crippen Method
logp	1.765		Crippen Method
mcvol	128.660	ml/mol	McGowan Method
pc	3191.93	kPa	Joback Method
rinpol	1216.00		NIST Webbook
rinpol	1216.00		NIST Webbook
tb	506.11	K	Joback Method
tc	715.74	K	Joback Method
tf	287.60	K	Joback Method
vc	0.483	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	245.54	J/mol×K	506.11	Joback Method
cpg	255.51	J/mol×K	541.05	Joback Method
cpg	264.99	J/mol×K	575.99	Joback Method
cpg	274.01	J/mol×K	610.93	Joback Method
cpg	282.57	J/mol×K	645.86	Joback Method
cpg	290.68	J/mol×K	680.80	Joback Method
cpg	298.35	J/mol×K	715.74	Joback Method
dvisc	0.0035969	Paxs	287.60	Joback Method

dvisc	0.0019254	Paxs	324.02	Joback Method
dvisc	0.0011694	Paxs	360.44	Joback Method
dvisc	0.0007783	Paxs	396.86	Joback Method
dvisc	0.0005547	Paxs	433.27	Joback Method
dvisc	0.0004167	Paxs	469.69	Joback Method
dvisc	0.0003261	Paxs	506.11	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406237&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406237&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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