

# Propanoic acid, 2-methyl, 2-iodoethyl ester

<b>Inchi:</b>	InChI=1S/C6H11IO2/c1-5(2)6(8)9-4-3-7/h5H,3-4H2,1-2H3
<b>InchiKey:</b>	BFYJFIMQAZKDGX-UHFFFAOYSA-N
<b>Formula:</b>	C6H11IO2
<b>SMILES:</b>	CC(C)C(=O)OCCI
<b>Mol. weight [g/mol]:</b>	242.05

## Physical Properties

Property code	Value	Unit	Source
gf	-178.60	kJ/mol	Joback Method
hf	-340.38	kJ/mol	Joback Method
hfus	14.97	kJ/mol	Joback Method
hvap	47.09	kJ/mol	Joback Method
log10ws	-1.90		Crippen Method
logp	1.621		Crippen Method
mcvol	128.660	ml/mol	McGowan Method
pc	3220.98	kPa	Joback Method
tb	505.67	K	Joback Method
tc	720.10	K	Joback Method
tf	272.60	K	Joback Method
vc	0.477	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	245.84	J/molxK	505.67	Joback Method
cpg	292.19	J/molxK	684.36	Joback Method
cpg	283.90	J/molxK	648.62	Joback Method
cpg	275.13	J/molxK	612.88	Joback Method
cpg	265.87	J/molxK	577.15	Joback Method
cpg	256.11	J/molxK	541.41	Joback Method
cpg	300.01	J/molxK	720.10	Joback Method
dvisc	0.0003082	Paxs	505.67	Joback Method
dvisc	0.0004035	Paxs	466.83	Joback Method
dvisc	0.0005547	Paxs	427.98	Joback Method

dvisc	0.0008126	Paxs	389.13	Joback Method
dvisc	0.0012956	Paxs	350.29	Joback Method
dvisc	0.0023206	Paxs	311.45	Joback Method
dvisc	0.0049075	Paxs	272.60	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R20085&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R20085&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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