

# 1H-Pyrrole-2,5-dione, 3-ethenyl-4-methyl-

<b>Other names:</b>	Maleimide, 2-methyl-3-vinyl- Methylvinylmaleimide 2-Methyl-2-vinylmaleimide 3-Methyl-4-vinyl-1H-pyrrole-2,5-dione
<b>Inchi:</b>	InChI=1S/C7H7NO2/c1-3-5-4(2)6(9)8-7(5)10/h3H,1H2,2H3,(H,8,9,10)
<b>InchiKey:</b>	LMEHYWBHZXZJNJ-UHFFFAOYSA-N
<b>Formula:</b>	C7H7NO2
<b>SMILES:</b>	<chem>C=CC1=C(C)C(O)=NC1=O</chem>
<b>Mol. weight [g/mol]:</b>	137.14
<b>CAS:</b>	21494-57-5

## Physical Properties

Property code	Value	Unit	Source
gf	28.56	kJ/mol	Joback Method
hf	-119.37	kJ/mol	Joback Method
hfus	15.48	kJ/mol	Joback Method
hvap	60.78	kJ/mol	Joback Method
log10ws	-1.11		Crippen Method
logp	0.986		Crippen Method
mcvol	103.150	ml/mol	McGowan Method
pc	4565.38	kPa	Joback Method
rinpol	1261.30		NIST Webbook
rinpol	1261.30		NIST Webbook
tb	603.15	K	Joback Method
tc	823.38	K	Joback Method
tf	421.69	K	Joback Method
vc	0.398	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	251.65	J/mol×K	603.15	Joback Method
cpg	261.63	J/mol×K	639.85	Joback Method
cpg	271.11	J/mol×K	676.56	Joback Method

cpg	280.05	J/mol×K	713.26	Joback Method
cpg	288.45	J/mol×K	749.97	Joback Method
cpg	296.27	J/mol×K	786.67	Joback Method
cpg	303.50	J/mol×K	823.38	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C21494575&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C21494575&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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