

# Ether, chloromethyl dichloromethyl

<b>Inchi:</b>	InChI=1S/C2H3Cl3O/c3-1-6-2(4)5/h2H,1H2
<b>InchiKey:</b>	MCXRKNYHEMVMIT-UHFFFAOYSA-N
<b>Formula:</b>	C2H3Cl3O
<b>SMILES:</b>	CICOC(Cl)Cl
<b>Mol. weight [g/mol]:</b>	149.40
<b>CAS:</b>	2799-32-8

## Physical Properties

Property code	Value	Unit	Source
gf	-177.27	kJ/mol	Joback Method
hf	-269.33	kJ/mol	Joback Method
hfus	11.19	kJ/mol	Joback Method
hvap	35.22	kJ/mol	Joback Method
log10ws	-1.80		Crippen Method
logp	1.960		Crippen Method
mvol	81.630	ml/mol	McGowan Method
pc	4271.86	kPa	Joback Method
tb	379.43	K	Joback Method
tc	576.92	K	Joback Method
tf	209.29	K	Joback Method
vc	0.306	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	116.71	J/molxK	379.43	Joback Method
cpg	120.63	J/molxK	412.34	Joback Method
cpg	124.42	J/molxK	445.26	Joback Method
cpg	128.09	J/molxK	478.17	Joback Method
cpg	131.62	J/molxK	511.09	Joback Method
cpg	135.02	J/molxK	544.00	Joback Method
cpg	138.28	J/molxK	576.92	Joback Method
dvisc	0.0043724	Paxs	209.29	Joback Method
dvisc	0.0022406	Paxs	237.65	Joback Method

dvisc	0.0013241	Paxs	266.00	Joback Method
dvisc	0.0008659	Paxs	294.36	Joback Method
dvisc	0.0006102	Paxs	322.72	Joback Method
dvisc	0.0004550	Paxs	351.07	Joback Method
dvisc	0.0003545	Paxs	379.43	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2799328&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2799328&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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