

# Methanesulfonic acid, ethyl ester

<b>Other names:</b>	Ethyl mesylate Ethyl methanesulfonate EMS NSC 26805 CB 1528 Ethyl ester of methanesulfonic acid Ethyl ester of methylsulfonic acid Half-Myleran Methylsulfonic acid, ethyl ester ENT 26396 Ethylester kyseliny methansulfonove Ethyl ester of methanesulphonic acid Ethyl ester of methylsulphonic acid Ethyl methanesulphonate Ethyl methansulfonate Ethyl methansulphonate Methanesulphonic acid ethyl ester Rcra waste number U119
<b>Inchi:</b>	InChI=1S/C3H8O3S/c1-3-6-7(2,4)5/h3H2,1-2H3
<b>InchiKey:</b>	PLUBXMRUUVWRLT-UHFFFAOYSA-N
<b>Formula:</b>	C3H8O3S
<b>SMILES:</b>	CCOS(C)(=O)=O
<b>Mol. weight [g/mol]:</b>	124.16
<b>CAS:</b>	62-50-0

## Physical Properties

Property code	Value	Unit	Source
gf	-599.16	kJ/mol	Joback Method
hf	-690.82	kJ/mol	Joback Method
hfus	16.09	kJ/mol	Joback Method
hvap	43.32	kJ/mol	Joback Method
log10ws	0.01		Crippen Method
logp	-0.017		Crippen Method
mcvol	87.090	ml/mol	McGowan Method
pc	5343.52	kPa	Joback Method
rinpol	146.35		NIST Webbook
tb	338.24	K	Joback Method

tc	501.95	K	Joback Method
tf	184.36	K	Joback Method
vc	0.347	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	147.22	J/mol×K	338.24	Joback Method
cpg	154.06	J/mol×K	365.53	Joback Method
cpg	160.80	J/mol×K	392.81	Joback Method
cpg	167.44	J/mol×K	420.10	Joback Method
cpg	173.95	J/mol×K	447.38	Joback Method
cpg	180.33	J/mol×K	474.67	Joback Method
cpg	186.57	J/mol×K	501.95	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C62500&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C62500&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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