

# Ethyl fluoroformate

<b>Other names:</b>	FCO2C2H5
<b>Inchi:</b>	InChI=1S/C3H5FO2/c1-2-6-3(4)5/h2H2,1H3
<b>InchiKey:</b>	PWEHMQBAMSTRDF-UHFFFAOYSA-N
<b>Formula:</b>	C3H5FO2
<b>SMILES:</b>	CCOC(=O)F
<b>Mol. weight [g/mol]:</b>	92.07
<b>CAS:</b>	461-64-3

## Physical Properties

Property code	Value	Unit	Source
affp	757.00	kJ/mol	NIST Webbook
basg	726.00	kJ/mol	NIST Webbook
gf	-454.35	kJ/mol	Joback Method
hf	-546.16	kJ/mol	Joback Method
hfus	9.39	kJ/mol	Joback Method
hvap	30.61	kJ/mol	Joback Method
log10ws	-0.77		Crippen Method
logp	1.112		Crippen Method
mcvol	62.340	ml/mol	McGowan Method
pc	4414.96	kPa	Joback Method
tb	329.50 ± 3.00	K	NIST Webbook
tc	513.10	K	Joback Method
tf	196.32	K	Joback Method
vc	0.245	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	106.62	J/molxK	343.60	Joback Method
cpg	111.64	J/molxK	371.85	Joback Method
cpg	116.56	J/molxK	400.10	Joback Method
cpg	121.38	J/molxK	428.35	Joback Method
cpg	126.08	J/molxK	456.60	Joback Method
cpg	130.67	J/molxK	484.85	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C461643&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C461643&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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