

# Methyl 1,1-dichloroethyl ether

Inchi:	InChI=1S/C3H6Cl2O/c1-3(4,5)6-2/h1-2H3
InchiKey:	ASXWDPUCYBGLRG-UHFFFAOYSA-N
Formula:	C3H6Cl2O
SMILES:	COC(C)(Cl)Cl
Mol. weight [g/mol]:	128.99

## Physical Properties

Property code	Value	Unit	Source
gf	-151.64	kJ/mol	Joback Method
hf	-277.70	kJ/mol	Joback Method
hfus	5.69	kJ/mol	Joback Method
hvap	32.16	kJ/mol	Joback Method
log10ws	-1.57		Crippen Method
logp	1.784		Crippen Method
mcvol	83.480	ml/mol	McGowan Method
pc	4015.93	kPa	Joback Method
rinpola	684.00		NIST Webbook
tb	362.09	K	Joback Method
tc	558.03	K	Joback Method
tf	208.06	K	Joback Method
vc	0.308	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	130.38	J/mol×K	362.09	Joback Method
cpg	160.88	J/mol×K	525.37	Joback Method
cpg	155.45	J/mol×K	492.72	Joback Method
cpg	149.71	J/mol×K	460.06	Joback Method
cpg	143.62	J/mol×K	427.40	Joback Method
cpg	137.18	J/mol×K	394.75	Joback Method
cpg	165.98	J/mol×K	558.03	Joback Method
dvisc	0.0003700	Paxs	362.09	Joback Method
dvisc	0.0004864	Paxs	336.42	Joback Method

dvisc	0.0006691	Paxs	310.75	Joback Method
dvisc	0.0009746	Paxs	285.08	Joback Method
dvisc	0.0015296	Paxs	259.40	Joback Method
dvisc	0.0026502	Paxs	233.73	Joback Method
dvisc	0.0052591	Paxs	208.06	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R629271&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R629271&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/98-745-7/Methyl-1-1-dichloroethyl-ether.pdf>

Generated by Cheméo on 2024-04-27 10:38:20.205882432 +0000 UTC m=+16503549.126459748.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.