

# 6-Undecanone, 5,7-dichloro (RS, SR)

<b>Inchi:</b>	InChI=1S/C11H20Cl2O/c1-3-5-7-9(12)11(14)10(13)8-6-4-2/h9-10H,3-8H2,1-2H3
<b>InchiKey:</b>	FJFNPTNEBBJNPW-UHFFFAOYSA-N
<b>Formula:</b>	C11H20Cl2O
<b>SMILES:</b>	CCCCC(Cl)C(=O)C(Cl)CCCC
<b>Mol. weight [g/mol]:</b>	239.18

## Physical Properties

Property code	Value	Unit	Source
gf	-115.92	kJ/mol	Joback Method
hf	-424.99	kJ/mol	Joback Method
hfus	27.19	kJ/mol	Joback Method
hvap	54.82	kJ/mol	Joback Method
log10ws	-4.24		Crippen Method
logp	4.151		Crippen Method
mvol	191.900	ml/mol	McGowan Method
pc	1942.37	kPa	Joback Method
rinpol	1460.00		NIST Webbook
rinpol	1460.00		NIST Webbook
tb	578.93	K	Joback Method
tc	767.04	K	Joback Method
tf	293.50	K	Joback Method
vc	0.744	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	451.00	J/molxK	578.93	Joback Method
cpg	465.56	J/molxK	610.28	Joback Method
cpg	479.40	J/molxK	641.63	Joback Method
cpg	492.53	J/molxK	672.99	Joback Method
cpg	504.99	J/molxK	704.34	Joback Method
cpg	516.79	J/molxK	735.69	Joback Method
cpg	527.95	J/molxK	767.04	Joback Method
dvisc	0.0060355	Paxs	293.50	Joback Method

dvisc	0.0023214	Paxs	341.07	Joback Method
dvisc	0.0011281	Paxs	388.64	Joback Method
dvisc	0.0006417	Paxs	436.22	Joback Method
dvisc	0.0004078	Paxs	483.79	Joback Method
dvisc	0.0002811	Paxs	531.36	Joback Method
dvisc	0.0002060	Paxs	578.93	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R630667&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R630667&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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