

# (3-Nitrophenyl) methanol, ethyl ether

<b>Inchi:</b>	InChI=1S/C9H11NO3/c1-2-13-7-8-4-3-5-9(6-8)10(11)12/h3-6H,2,7H2,1H3
<b>InchiKey:</b>	VPUMBFVYZSOPAJ-UHFFFAOYSA-N
<b>Formula:</b>	C9H11NO3
<b>SMILES:</b>	CCOCc1cccc([N+](=O)[O-])c1
<b>Mol. weight [g/mol]:</b>	181.19

## Physical Properties

Property code	Value	Unit	Source
gf	58.23	kJ/mol	Joback Method
hf	-147.01	kJ/mol	Joback Method
hfus	25.27	kJ/mol	Joback Method
hvap	57.57	kJ/mol	Joback Method
log10ws	-2.93		Crippen Method
logp	2.131		Crippen Method
mcvol	137.200	ml/mol	McGowan Method
pc	3188.33	kPa	Joback Method
rinpol	1509.00		NIST Webbook
rinpol	1509.00		NIST Webbook
tb	611.24	K	Joback Method
tc	847.17	K	Joback Method
tf	395.97	K	Joback Method
vc	0.531	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	334.13	J/mol×K	611.24	Joback Method
cpg	347.05	J/mol×K	650.56	Joback Method
cpg	359.12	J/mol×K	689.88	Joback Method
cpg	370.35	J/mol×K	729.20	Joback Method
cpg	380.77	J/mol×K	768.52	Joback Method
cpg	390.39	J/mol×K	807.85	Joback Method
cpg	399.25	J/mol×K	847.17	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U374644&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U374644&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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