

# Benzamide, 2-chloro-N-ethyl-

<b>Inchi:</b>	InChI=1S/C9H10ClNO/c1-2-11-9(12)7-5-3-4-6-8(7)10/h3-6H,2H2,1H3,(H,11,12)
<b>InchiKey:</b>	OUMQNCAKBPDIY-UHFFFAOYSA-N
<b>Formula:</b>	C9H10ClNO
<b>SMILES:</b>	CCN=C(O)c1ccccc1Cl
<b>Mol. weight [g/mol]:</b>	183.63

## Physical Properties

Property code	Value	Unit	Source
hf	-99.57	kJ/mol	Joback Method
hvap	63.02	kJ/mol	Joback Method
log10ws	-2.46		Crippen Method
logp	2.665		Crippen Method
mcvol	137.700	ml/mol	McGowan Method
pc	3049.04	kPa	Joback Method
rinsol	1571.00		NIST Webbook
rinsol	1571.00		NIST Webbook
tb	643.15	K	Joback Method
tc	862.26	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U407427&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U407427&amp;Units=SI</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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