

# Carbonic acid, monoamide, N-dodecyl-, decyl ester

Inchi:	InChI=1S/C23H47NO2/c1-3-5-7-9-11-13-14-15-17-19-21-24-23(25)26-22-20-18-16-12-10
InchiKey:	YDPXXVRDIFBDLK-UHFFFAOYSA-N
Formula:	C23H47NO2
SMILES:	CCCCCCCCCCCCN=C(O)CCCCCCCCCCC
Mol. weight [g/mol]:	369.62

## Physical Properties

Property code	Value	Unit	Source
hf	-730.07	kJ/mol	Joback Method
hvap	89.27	kJ/mol	Joback Method
log10ws	-8.02		Crippen Method
logp	7.979		Crippen Method
mcvol	352.350	ml/mol	McGowan Method
pc	839.67	kPa	Joback Method
rinpol	3128.00		NIST Webbook
rinpol	3128.00		NIST Webbook
tb	916.80	K	Joback Method
tc	1126.12	K	Joback Method

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U415187&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U415187&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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