

# N-(Alpha-naphthyl)-N',N'-diphenyl urea

<b>Inchi:</b>	InChI=1S/C23H18N2O/c26-23(24-22-17-9-11-18-10-7-8-16-21(18)22)25(19-12-3-1-4-13
<b>InchiKey:</b>	XRAPDLHFGPTBQZ-UHFFFAOYSA-N
<b>Formula:</b>	C23H18N2O
<b>SMILES:</b>	OC(=Nc1cccc2ccccc12)N(c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	338.40
<b>CAS:</b>	60302-02-5

## Physical Properties

Property code	Value	Unit	Source
hf	358.87	kJ/mol	Joback Method
hvap	98.04	kJ/mol	Joback Method
log10ws	-6.67		Crippen Method
logp	6.224		Crippen Method
mcvol	265.720	ml/mol	McGowan Method
pc	2041.91	kPa	Joback Method
tb	1010.82	K	Joback Method
tc	1269.49	K	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C60302025&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C60302025&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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