

# Glutaric acid, 3-chlorophenyl 2-methoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C18H17ClO5/c1-22-15-8-2-3-9-16(15)24-18(21)11-5-10-17(20)23-14-7-4-6-13
<b>InchiKey:</b>	CXHYOVDNXYIHBH-UHFFFAOYSA-N
<b>Formula:</b>	C18H17ClO5
<b>SMILES:</b>	COc1ccccc1OC(=O)CCCC(=O)Oc1cccc(Cl)c1
<b>Mol. weight [g/mol]:</b>	348.78

## Physical Properties

Property code	Value	Unit	Source
gf	-278.53	kJ/mol	Joback Method
hf	-602.29	kJ/mol	Joback Method
hfus	40.64	kJ/mol	Joback Method
hvap	86.65	kJ/mol	Joback Method
log10ws	-4.97		Crippen Method
logp	4.030		Crippen Method
mcvol	249.950	ml/mol	McGowan Method
pc	1930.44	kPa	Joback Method
rinpol	2700.00		NIST Webbook
rinpol	2700.00		NIST Webbook
tb	886.99	K	Joback Method
tc	1117.53	K	Joback Method
tf	566.97	K	Joback Method
vc	0.943	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	715.76	J/molxK	886.99	Joback Method
cpg	761.53	J/molxK	1079.11	Joback Method
cpg	754.98	J/molxK	1040.69	Joback Method
cpg	747.15	J/molxK	1002.26	Joback Method
cpg	738.00	J/molxK	963.84	Joback Method
cpg	727.55	J/molxK	925.41	Joback Method
cpg	766.80	J/molxK	1117.53	Joback Method
dvisc	0.0000491	Paxs	886.99	Joback Method

dvisc	0.0000610	Paxs	833.65	Joback Method
dvisc	0.0000779	Paxs	780.32	Joback Method
dvisc	0.0001033	Paxs	726.98	Joback Method
dvisc	0.0001433	Paxs	673.64	Joback Method
dvisc	0.0002101	Paxs	620.31	Joback Method
dvisc	0.0003311	Paxs	566.97	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391764&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391764&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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