

# Isophthalic acid, monoamide, N-butyl-, isoheptyl ester

Inchi:	lnChI=1S/C18H27NO3/c1-4-5-11-19-17(20)15-9-6-10-16(13-15)18(21)22-12-7-8-14(2)3/
InchiKey:	IXGZPTZANLUORU-UHFFFAOYSA-N
Formula:	C18H27NO3
SMILES:	CCCCN=C(O)c1ccccc(C(=O)OCCCC(C)C)c1
Mol. weight [g/mol]:	305.41

## Physical Properties

Property code	Value	Unit	Source
hf	-519.67	kJ/mol	Joback Method
hvap	87.44	kJ/mol	Joback Method
log10ws	-4.71		Crippen Method
logp	4.384		Crippen Method
mcvol	259.710	ml/mol	McGowan Method
pc	1497.67	kPa	Joback Method
rinpol	2538.00		NIST Webbook
rinpol	2538.00		NIST Webbook
tb	887.49	K	Joback Method
tc	1095.76	K	Joback Method

## Sources

NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U345811&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U345811&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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