

I-Leucine, N-neopentylloxycarbonyl-N-methyl-, hexyl ester

Inchi:	InChI=1S/C19H37NO4/c1-8-9-10-11-12-23-17(21)16(13-15(2)3)20(7)18(22)24-14-19(4,5
InchiKey:	RIJMSNAZBLHAIO-UHFFFAOYSA-N
Formula:	C19H37NO4
SMILES:	CCCCCOC(=O)C(CC(C)C)N(C)C(=O)OCC(C)(C)C
Mol. weight [g/mol]:	343.50

Physical Properties

Property code	Value	Unit	Source
gf	-250.00	kJ/mol	Joback Method
hf	-876.87	kJ/mol	Joback Method
hfus	39.10	kJ/mol	Joback Method
hvap	76.17	kJ/mol	Joback Method
log10ws	-4.68		Crippen Method
logp	4.639		Crippen Method
mcvol	303.430	ml/mol	McGowan Method
pc	1175.24	kPa	Joback Method
rinsol	2006.00		NIST Webbook
tb	795.03	K	Joback Method
tc	982.13	K	Joback Method
tf	453.10	K	Joback Method
vc	1.143	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	953.99	J/molxK	795.03	Joback Method
cpg	972.25	J/molxK	826.21	Joback Method
cpg	989.43	J/molxK	857.40	Joback Method
cpg	1005.55	J/molxK	888.58	Joback Method
cpg	1020.67	J/molxK	919.76	Joback Method
cpg	1034.81	J/molxK	950.95	Joback Method
cpg	1048.01	J/molxK	982.13	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U321910&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/99-614-1/l-Leucine-N-neopentylloxycarbonyl-N-methyl-hexyl-ester.pdf>

Generated by Cheméo on 2024-04-25 19:01:20.590096758 +0000 UTC m=+16360929.510674073.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.