

# DL-Valine, N-methyl-N-octyloxycarbonyl-, butyl ester

<b>Inchi:</b>	InChI=1S/C19H37NO4/c1-6-8-10-11-12-13-15-24-19(22)20(5)17(16(3)4)18(21)23-14-9-7
<b>InchiKey:</b>	FHYFFGWGOVKLII-UHFFFAOYSA-N
<b>Formula:</b>	C19H37NO4
<b>SMILES:</b>	CCCCCCCCOC(=O)N(C)C(C(=O)OCCCC)C(C)C
<b>Mol. weight [g/mol]:</b>	343.50

## Physical Properties

Property code	Value	Unit	Source
gf	-252.84	kJ/mol	Joback Method
hf	-868.12	kJ/mol	Joback Method
hfus	46.52	kJ/mol	Joback Method
hvap	77.47	kJ/mol	Joback Method
log10ws	-4.92		Crippen Method
logp	4.783		Crippen Method
mvol	303.430	ml/mol	McGowan Method
pc	1160.08	kPa	Joback Method
rinpol	2143.00		NIST Webbook
rinpol	2143.00		NIST Webbook
tb	798.26	K	Joback Method
tc	982.17	K	Joback Method
tf	450.68	K	Joback Method
vc	1.153	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	952.86	J/molxK	798.26	Joback Method
cpg	971.04	J/molxK	828.91	Joback Method
cpg	988.16	J/molxK	859.56	Joback Method
cpg	1004.24	J/molxK	890.22	Joback Method
cpg	1019.32	J/molxK	920.87	Joback Method
cpg	1033.40	J/molxK	951.52	Joback Method
cpg	1046.51	J/molxK	982.17	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U392900&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U392900&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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