

# Urea, triphenyl-

<b>Other names:</b>	Triphenylurea
<b>Inchi:</b>	InChI=1S/C19H16N2O/c22-19(20-16-10-4-1-5-11-16)21(17-12-6-2-7-13-17)18-14-8-3-9-
<b>InchiKey:</b>	HKQGDIKHDDGEOE-UHFFFAOYSA-N
<b>Formula:</b>	C19H16N2O
<b>SMILES:</b>	OC(=Nc1ccccc1)N(c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	288.34
<b>CAS:</b>	5663-04-7

## Physical Properties

Property code	Value	Unit	Source
hf	261.83	kJ/mol	Joback Method
hvap	86.83	kJ/mol	Joback Method
log10ws	-4.87		Crippen Method
logp	5.070		Crippen Method
mcvol	228.820	ml/mol	McGowan Method
pc	2358.78	kPa	Joback Method
tb	895.34	K	Joback Method
tc	1145.94	K	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5663047&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5663047&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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