

# Glutaric acid, 3-chlorophenyl 5-methyl-2-methoxybenzyl ester

<b>Inchi:</b>	InChI=1S/C19H19ClO5/c1-13-9-10-16(23-2)17(11-13)25-19(22)8-4-7-18(21)24-15-6-3-5
<b>InchiKey:</b>	GGOJLNLPHTXSHL-UHFFFAOYSA-N
<b>Formula:</b>	C19H19ClO5
<b>SMILES:</b>	COc1ccc(C)cc1OC(=O)CCCC(=O)Oc1cccc(Cl)c1
<b>Mol. weight [g/mol]:</b>	362.80

## Physical Properties

Property code	Value	Unit	Source
gf	-279.74	kJ/mol	Joback Method
hf	-634.40	kJ/mol	Joback Method
hfus	42.84	kJ/mol	Joback Method
hvap	89.53	kJ/mol	Joback Method
log10ws	-5.45		Crippen Method
logp	4.338		Crippen Method
mvol	264.040	ml/mol	McGowan Method
pc	1755.07	kPa	Joback Method
rinpol	2776.00		NIST Webbook
rinpol	2776.00		NIST Webbook
tb	914.85	K	Joback Method
tc	1144.37	K	Joback Method
tf	590.76	K	Joback Method
vc	0.999	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	770.95	J/molxK	914.85	Joback Method
cpg	815.92	J/molxK	1106.12	Joback Method
cpg	809.65	J/molxK	1067.86	Joback Method
cpg	802.03	J/molxK	1029.61	Joback Method
cpg	793.04	J/molxK	991.36	Joback Method
cpg	782.68	J/molxK	953.10	Joback Method
cpg	820.83	J/molxK	1144.37	Joback Method
dvisc	0.0000435	Paxs	914.85	Joback Method

dvisc	0.0000536	Paxs	860.83	Joback Method
dvisc	0.0000679	Paxs	806.82	Joback Method
dvisc	0.0000889	Paxs	752.80	Joback Method
dvisc	0.0001216	Paxs	698.79	Joback Method
dvisc	0.0001751	Paxs	644.77	Joback Method
dvisc	0.0002695	Paxs	590.76	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U393930&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U393930&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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