

# Diethylmalonic acid, 3-bromobenzyl pentyl ester

Inchi:	InChI=1S/C19H27BrO4/c1-4-7-8-12-23-17(21)19(5-2,6-3)18(22)24-14-15-10-9-11-16(20)
InchiKey:	COYLCBQWRSRLMJ-UHFFFAOYSA-N
Formula:	C19H27BrO4
SMILES:	CCCCCOC(=O)C(CC)(CC)C(=O)OCc1cccc(Br)c1
Mol. weight [g/mol]:	399.32

## Physical Properties

Property code	Value	Unit	Source
gf	-238.80	kJ/mol	Joback Method
hf	-682.45	kJ/mol	Joback Method
hfus	42.06	kJ/mol	Joback Method
hvap	84.28	kJ/mol	Joback Method
log10ws	-6.02		Crippen Method
logp	5.032		Crippen Method
mcvol	287.190	ml/mol	McGowan Method
pc	1542.71	kPa	Joback Method
rinpol	2345.00		NIST Webbook
rinpol	2345.00		NIST Webbook
tb	881.29	K	Joback Method
tc	1097.30	K	Joback Method
tf	549.37	K	Joback Method
vc	1.091	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	860.19	J/molxK	881.29	Joback Method
cpg	921.69	J/molxK	1061.30	Joback Method
cpg	911.37	J/molxK	1025.30	Joback Method
cpg	900.12	J/molxK	989.30	Joback Method
cpg	887.87	J/molxK	953.29	Joback Method
cpg	874.58	J/molxK	917.29	Joback Method
cpg	931.13	J/molxK	1097.30	Joback Method
dvisc	0.0000389	Paxs	881.29	Joback Method

dvisc	0.0000501	Paxs	825.97	Joback Method
dvisc	0.0000670	Paxs	770.65	Joback Method
dvisc	0.0000937	Paxs	715.33	Joback Method
dvisc	0.0001387	Paxs	660.01	Joback Method
dvisc	0.0002204	Paxs	604.69	Joback Method
dvisc	0.0003847	Paxs	549.37	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U368413&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U368413&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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