

# Benzeneacetic acid, 4-chloro-, undec-2-en-1-yl ester

Inchi:	InChI=1S/C19H27ClO2/c1-2-3-4-5-6-7-8-9-10-15-22-19(21)16-17-11-13-18(20)14-12-17
InchiKey:	XCNFLAOJPRZLSG-MDZDMXLPSA-N
Formula:	C19H27ClO2
SMILES:	CCCCCCCC=CCOC(=O)Cc1ccc(Cl)cc1
Mol. weight [g/mol]:	322.87

## Physical Properties

Property code	Value	Unit	Source
gf	46.25	kJ/mol	Joback Method
hf	-353.75	kJ/mol	Joback Method
hfus	45.80	kJ/mol	Joback Method
hvap	74.33	kJ/mol	Joback Method
log10ws	-6.28		Crippen Method
logp	5.732		Crippen Method
mcvol	270.190	ml/mol	McGowan Method
pc	1408.01	kPa	Joback Method
rinsol	2361.00		NIST Webbook
tb	783.66	K	Joback Method
tc	985.06	K	Joback Method
tf	439.83	K	Joback Method
vc	1.044	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	773.81	J/molxK	783.66	Joback Method
cpg	846.24	J/molxK	951.49	Joback Method
cpg	833.55	J/molxK	917.92	Joback Method
cpg	820.00	J/molxK	884.36	Joback Method
cpg	805.56	J/molxK	850.79	Joback Method
cpg	790.18	J/molxK	817.23	Joback Method
cpg	858.15	J/molxK	985.06	Joback Method
dvisc	0.0000652	Paxs	783.66	Joback Method
dvisc	0.0000847	Paxs	726.36	Joback Method

dvisc	0.0001150	Paxs	669.05	Joback Method
dvisc	0.0001655	Paxs	611.75	Joback Method
dvisc	0.0002566	Paxs	554.44	Joback Method
dvisc	0.0004404	Paxs	497.14	Joback Method
dvisc	0.0008699	Paxs	439.83	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U407002&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U407002&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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