

# 1,5-Diisocyanatonaphthalene

<b>Inchi:</b>	InChI=1S/C12H6N2O2/c15-7-13-11-5-1-3-9-10(11)4-2-6-12(9)14-8-16/h1-6H
<b>InchiKey:</b>	SBJCUZQNHOLYMD-UHFFFAOYSA-N
<b>Formula:</b>	C12H6N2O2
<b>SMILES:</b>	O=C=Nc1cccc2c(N=C=O)cccc12
<b>Mol. weight [g/mol]:</b>	210.19

## Physical Properties

Property code	Value	Unit	Source
hf	102.83	kJ/mol	Joback Method
hvap	66.61	kJ/mol	Joback Method
log10ws	-12.24		Crippen Method
logp	2.774		Crippen Method
mcvol	151.220	ml/mol	McGowan Method
pc	3560.02	kPa	Joback Method
tb	662.92	K	Joback Method
tc	905.69	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.chemic.org/files/research/kdb/mol/mol1492.mol">https://www.chemic.org/files/research/kdb/mol/mol1492.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

**mcvol:** McGowan's characteristic volume  
**pc:** Critical Pressure  
**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature

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